

Product Data Sheet

DuPont[™] AmberLite[™] 14i Inert Resin

Polypropylene, Inert Resin for Industrial Demineralization Applications

Description DuPont[™] AmberLite[™] 14i Inert Resin is a floating, non-functionalized, transparent, cylindrical-shaped resin specifically designed for use as an upper layer in down-flow regenerated ion exchange systems, such as floating beds. This inert resin has a specific gravity lower than water, which ensures it will stay above the ion exchange resin bed. The inert forms a protective layer to prevent plugging of the distribution nozzles during the compaction/bed-lift step in case fines are present in the resin bed.

Use of AmberLite[™] 14i is optional in AMBERPACK[™] Ion Exchange Systems with AmberLite[™] HPR packed bed resins installed.

Applications • Demineralization

- System Designs • Packed beds – for AMBERPACK™ Systems or other down-flow regenerated packed bed systems
 - Counter-current / Air hold-down

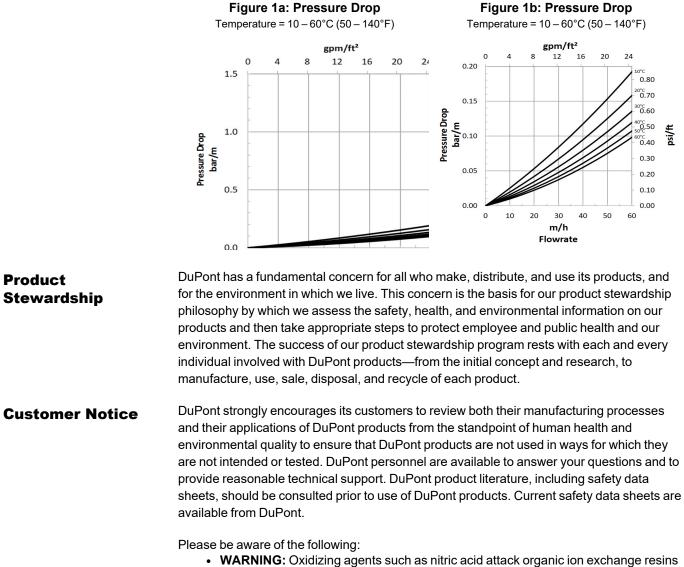
HistoricalAmberLite™ 14i Inert Resin has previously been sold as AmberLite™ RF14 InertReferenceResin.

Typical Properties	Physical Properties	
	Polymer	Polypropylene
	Туре	Inert
	Functional Group	None
	Physical Form	Colorless, translucent, cylinders
	Particle Size	
	Particle Diameter	1.2 – 1.5 mm
	Particle Length	1.3 – 1.7 mm
	Density	
	Particle Density	0.95 g/mL
	Shipping Weight	500 – 580 g/L
Suggested	Temperature Range	5–100°C (41–212°F)
Operating	pH Range	0 – 14
Conditions		

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for <u>separate beds</u> (Form No. 45-D01131-en) in water treatment, please refer to our Tech Fact.

Hydraulic Characteristics

Estimated pressure drop for DuPont[™] AmberLite[™] 14i Inert Resin as a function of service flowrate and temperature is shown in Figure 1a and a magnified scale of the same is shown in Figure 1b. These estimated pressure drop expectations are valid at the start of the service run with clean water and a well-classified bed.



• WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

www.dupont.com/water/contact-us

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