

Product Data Sheet

	DuPont[™] AmberLite[™] MB6113 H/OH Ion Exchange Resin Dyed Mixture of Gaussian, Gel, Strong Acid Cation and Strong Base Anion Exchange Resins for Industrial Demineralization Applications
Description	DuPont [™] AmberLite [™] MB6113 H/OH Ion Exchange Resin is a dyed mixture of strong acid cation and strong base anion exchange resins. It is fully regenerated, ready-to-use, non-regenerable, pre-mixed resin developed for the production of high-purity water. The pre-mixed resin also allows for faster initial rinse-up prior to service, which minimizes rinse wastewater volume at start-up.
	AmberLite [™] MB6113 H/OH is specifically designed for a full demineralization of water. A color indicator enables the visualization of the resin exhaustion point, which allows the production of demineralized water without a conductivity meter to detect the service end-point. After exhaustion, AmberLite [™] MB6113 H/OH cannot be regenerated.
	AmberLite™ MB6113 H/OH is the reference mixed bed for the production of demineralized water in small cartridge systems. The resin mixture is prepared from high-quality components and the proprietary manufacturing process ensures a homogeneous blue-green color consistently from batch to batch.
	In operation, the resin provides a stable, high-quality demineralized water for laboratories, steam ironing, or battery fill-up. The sharp visible color change from regenerated (blue-green) to exhausted (amber-yellow) makes AmberLite™ MB6113 H/OH the product of choice for small cartridge deionization systems.
Applications	Working mixed bed in cartridge applications
System Designs	Non-regenerated mixed beds
Historical Reference	AmberLite™ MB6113 H/OH lon Exchange Resin has previously been sold as AmberLite™ MB6113 Ion Exchange Resin.

Typical Properties

	Cation Resin	Anion Resin
Physical Properties		
Copolymer	Styrene-divinylbenzene	Styrene-divinylbenzene
Matrix	Gel	Gel
Туре	Strong acid cation	Strong base anion, Type I
Functional Group	Sulfonic acid	Trimethylammonium
Physical Form	Translucent, spherical beads	Translucent, spherical beads
	Blue-green (as delivered) to Amber-yellow (exhausted)	
Volume Ratio	37-46%	63-54%
Chemical Properties		
Ionic Form as Shipped	H⁺	OH⁻
Stability		
Whole Beads	≥90%	
Density		
Shipping Weight	700 g/L	

Product Performance

Operating Capacity

The operating capacity of DuPont[™] AmberLite[™] MB6113 H/OH Ion Exchange Resin can be estimated using the following formula, which gives an approximate determination of volume of water that can be treated:

$$BV = \frac{400}{TDS (meq/L)} \qquad \text{or} \qquad \frac{gal}{ft^3} = \frac{150000}{TDS (as ppm CaCO_3)}$$

where BV (Bed Volume) is the number of liters of a feedwater containing a TDS (Total Dissolved Solids) given in meq/L that can be demineralized with one liter of the resin mixture when run to exhaustion (or US gallons per cubic foot of the resin with TDS as ppm $CaCO_3$).

Treated Water Quality

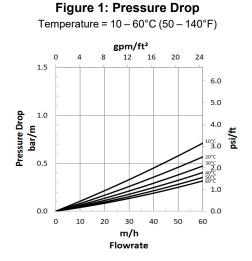
AmberLite[™] MB6113 H/OH Ion Exchange Resin provides a high-quality demineralized water with a conductivity < 1 µS/cm and neutral pH that will satisfy most of the cartridge and laboratory applications.

Suggested	Temperature Range (H+/OH ⁻ form) [‡]	5-60°C (41-140°F)
Operating Conditions	[‡] Operating mixed beds at elevated temperatures, for example above 60 – 70°C (140 – 158°F), may impact the purity of the loop and resin life. Contact our technical representative for details.	
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For additional information regarding recommended minimum bed depth and operating conditions for <u>mixed beds</u> (Form No. 45-D01127-en) in water treatment, please refer to our Tech Facts.

Hydraulic Characteristics

Estimated pressure drop for DuPont[™] AmberLite[™] MB6113 H/OH Ion Exchange Resin as a function of service flowrate and temperature is shown in Figure 1. These pressure drop expectations are valid at the start of the service run with clean water and a well-classified bed.



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Please be aware of the following:

• WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

Have a question? Contact us at:

www.dupont.com/water/contact-us

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